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12 REVIEW Solutions SECTION 2

SHORT ANSWER Answer the following questions in the space provided. 1. The following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas in the liquid.

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A homogeneous mixture that contains particles in a dispersed phase that do not settle out is a. solution. The solution process with water as the solvent. hydration. Liquid solutes and solvents that are not soluble in each other are. immiscible. A solute

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molecule that is surrounded by solvent molecules is solvated.

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Chapter 12 Solutions review. STUDY.
PLAY. solution. homogeneous mixture of two or more substances, consisting of ions

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or molecules. solute. gas or solid dissolved in
a liquid or solution component in small
amount. solvent. liquid that dissolves a gas
or solid or the solution component in
greater amount.

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CHAPTER 12 REVIEW Solutions
SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the solvent

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in each of the following examples:

~~mc06sec-cFMsr i-vi~~

Chapter 12 Review Solutions 1) 56 2) $= 4, = 10, = 16, = 22, = 28$ 3) $= 4, = 8, = , = 64, = 4)$
 $an = 2n - 3$ 5) $= \cdot 4n$ 6) $= 5, = 3, = 1, = -1$
7) $3 + 4 + 5 + \dots + (n + 2)$ 8) 9) 15 10) $d =$
 $-5; = 4, = -1, = -6, = -11$ 11) $= 94 - 10n; = 14$

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12) 220 13) = 73, d = -5, = - 5 14) 34980 15)
25

~~Chapter 12 Review Solutions, = 10, = 16, =
22, = 28 + 1) = 2.~~

CHAPTER . 12 . REVIEW ""-- Solutions .

SHORT ANSWER Answer the following
questions in the space provided. 1. Describe

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the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

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CHAPTER 12 REVIEW Solutions
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. _____

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b. Mary needs a 0.02 M solution of NaNO_3 . She calculates that she needs 2.00

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solutions: a worksheet that has chemistry
learners use words to describe them! This is
an outstanding exercise that will stretch

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learners to show what they know. They answer questions comparing solutions, suspensions, and colloids.

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Chapter 12 Review Problems

INSTRUCTIONS: Write your answers on

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lined or blank sheets of paper. You do not need to write the question, ONLY WRITE THE PROBLEM NUMBER and ANSWERS/SOLUTIONS. • For problems that involve calculations, you must show your work to get full credit.

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Chapter 12-Solutions Section 1- Types of

Mixtures Solutions are homogeneous

mixtures of two or more substances in a

single phase. Soluble describes a substance as

capable of being dissolved. Solvent is the

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dissolving medium in a solution.

Chemistry

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Answers Kora~~

Modern Chemistry Chapter 12-Solutions
Section 1- Types of Mixtures Solutions are
homogeneous mixtures of two or more
substances in a single phase. Soluble

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describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution. Solute is the substance that is dissolved in a solution.

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Exercise 18.

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